

An Introduction to Cathodic Protection for Underground Structures

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By

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An Introduction to Cathodic Protection



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CONTENTS

1. INTRODUCTION

- 1.1 Purpose
- 1.2 Corrosion
- 1.3 Cathodic protection
- 1.4 Types of cathodic protection systems

2. CATHODIC PROTECTION DESIGN

- 2.1 Required information
- 2.2 Determining type and design of cathodic protection system

3. CURRENT REQUIREMENT TESTING

- 3.1 Required current.
- 3.2 Sample test.

4. EXAMPLES OF GALVANIC CATHODIC PROTECTION DESIGN

5. EXAMPLES OF IMPRESSED CURRENT CATHODIC PROTECTION DESIGN

1. INTRODUCTION

1.1 Purpose. This course presents design guidance for cathodic protection systems.

1.2 Corrosion. Corrosion is an electrochemical process in which a current leaves a structure at the anode site, passes through an electrolyte, and re-enters the structure at the cathode site. For example, one small section of a pipeline may be anodic (positively charged) because it is in a soil with low resistivity compared to the rest of the line. Current would leave the pipeline at that anode site, pass through the soil, and re-enter the pipeline at a cathode (negatively charged) site. Current flows because of a potential difference between the anode and cathode. That is, the anode potential is more negative than the cathode potential, and this difference is the driving force for the corrosion current. The total system – anode, cathode, electrolyte, and metallic connection between anode and cathode is termed a *corrosion cell*.

1.3 Cathodic protection. Cathodic protection is a method to reduce corrosion by minimizing the difference in potential between anode and cathode. This is achieved by applying a current to the structure to be protected (such as a pipeline) from some outside source. When enough current is applied, the whole structure will be at one potential; thus, anode and cathode sites will not exist. Cathodic protection is commonly used on many types of structures, such as pipelines, underground storage tanks, locks, and ship hulls.

1.4 Types of cathodic protection systems. There are two main types of cathodic protection systems: galvanic and impressed current. Figure 1 shows these two types. Note that both types have anodes (from which current flows into the electrolyte), a continuous electrolyte from the anode to the protected structure, and an external metallic connection (wire). These items are essential for all cathodic protection systems.

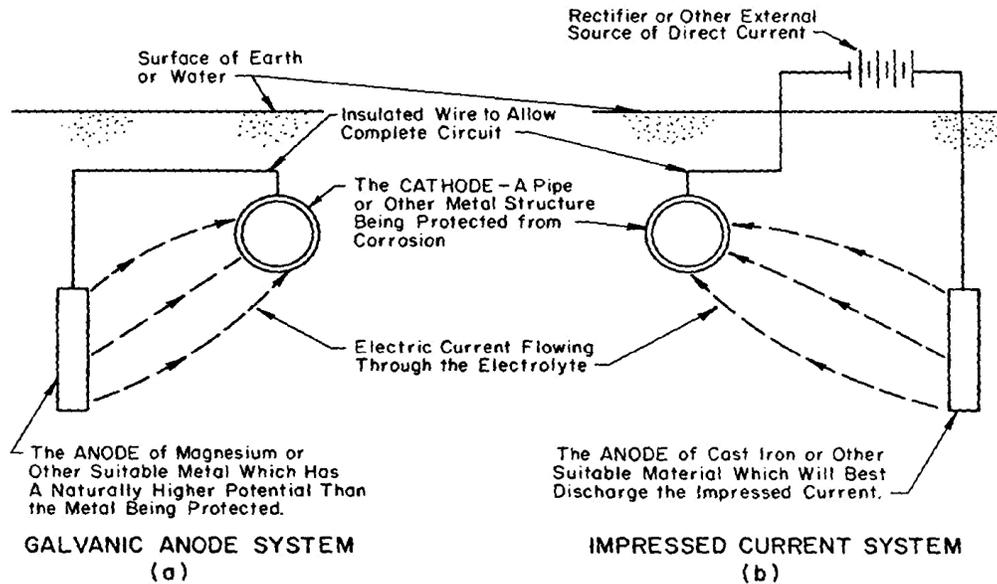


Figure 1

Galvanic (a) and impressed (b) current systems for cathodic protection

1.4.1 Galvanic system. A galvanic cathodic protection system makes use of the corrosive potentials for different metals. Without cathodic protection, one area of the structure exists at a more negative potential than another, and corrosion results. If, however, a much less inert object (that is, with much more negative potential, such as a magnesium anode) is placed adjacent to the structure to be protected, such as a pipeline, and a metallic connection (insulated wire) is installed between the object and the structure, the object will become the anode and the entire structure will become the cathode. That is, the new object corrodes sacrificially to protect the structure as shown in Figure 1(a). Thus, the galvanic cathode protection system is called a *sacrificial anode* cathodic protection system because the anode corrodes sacrificially to protect the structure. Galvanic anodes are usually made of either magnesium or zinc because of these metals' higher potential compared to steel structures.

1.4.2 Impressed current systems. Impressed current cathodic protection systems use the same elements as the galvanic protection system, only the structure is protected by applying a current to it from an anode. The anode and the structure are connected by an insulated wire, as for the galvanic system. Current flows from the

anode through the electrolyte onto the structure, just as in the galvanic system. The main difference between galvanic and impressed current systems is that the galvanic system relies on the difference in potential between the anode and the structure, whereas the impressed current system uses an external power source to drive the current, as shown in Figure 1(b). The external power source is usually a rectifier that changes input AC power to the proper DC power level. The rectifier can be adjusted so that proper output can be maintained during the system's life. Impressed current cathodic protection system anodes typically are high-silicone cast iron or graphite.

2. CATHODIC PROTECTION DESIGN

2.1 Required information. Before deciding which type, galvanic or impressed current, cathodic protection system will be used and before the system is designed, certain preliminary data must be gathered.

2.1.1 Physical dimensions of structure to be protected. One important element in designing a cathodic protection system is the structure's physical dimensions (for example, length, width, height, and diameter). These data are used to calculate the surface area to be protected.

2.1.2 Drawing of structure to be protected. The installation drawings must include sizes, shapes, material type, and locations of parts of the structure to be protected.

2.1.3 Electrical isolation. If a structure is to be protected by the cathodic system, it must be electrically connected to the anode, as Figure 1 shows. Sometimes parts of a structure or system are electrically isolated from each other by insulators. For example, in a gas pipeline distribution system, the inlet pipe to each building might contain an electric insulator to isolate in-house piping from the pipeline. Also, an electrical insulator might be used at a valve along the pipeline to electrically isolate one section of the system from another. Since each electrically isolated part of a structure would need its own cathodic protection, the locations of these insulators must be determined.

2.1.4 Short circuits. All short circuits must be eliminated from existing and new cathodic protection systems. A short circuit can occur when one pipe system contacts another, causing interference with the cathodic protection system. When updating existing systems, eliminating short circuits would be a necessary first step.

2.1.5 Corrosion history of structures in the area. Studying the corrosion history in the area can prove very helpful when designing a cathodic protection system. The study should reinforce predictions for corrosivity of a given structure and its environment; in addition, it may reveal abnormal conditions not otherwise suspected. Facilities personnel can be a good source of information for corrosion history.

2.1.6 Electrolyte resistivity survey. A structure's corrosion rate is proportional to the electrolyte resistivity. Without cathodic protection, as electrolyte resistivity decreases,

more current is allowed to flow from the structure into the electrolyte; thus, the structure corrodes more rapidly. As electrolyte resistivity increases, the corrosion rate decreases (Table 1). Resistivity can be measured either in a laboratory or at the site with the proper instruments. The resistivity data will be used to calculate the sizes of anodes and rectifier required in designing the cathodic protection system.

Soil resistivity range (ohm-cm)	Corrosivity
0 to 2000	Severe
2000 to 10,000	Moderate to severe
10,000 to 30,000	Mild
Above 30,000	Not likely

Table 1
Corrosivity of soils on steel based on soil resistivity

2.1.7 Electrolyte pH survey. Corrosion is also proportional to electrolyte pH. In general, steel's corrosion rate increases as pH decreases when soil resistivity remains constant.

2.1.8 Structure versus electrolyte potential survey. For existing structures, the potential between the structure and the electrolyte will give a direct indication of the corrosivity. According to NACE Standard No. RP-01, the potential requirement for cathodic protection is a negative (cathodic) potential of at least 0.85 volt as measured between the structure and a saturated copper-copper sulfate reference electrode in contact with the electrolyte. A potential which is less negative than -0.85 volt would probably be corrosive, with corrosivity increasing as the negative value decreases (becomes more positive).

2.1.9 Current requirement. A critical part of design calculations for cathodic protection systems on existing structures is the amount of current required per square foot (called *current density*) to change the structure's potential to -0.85 volt. The current density required to shift the potential indicates the structure's surface condition. A well

coated structure (for example, a pipeline well coated with coal-tar epoxy) will require a very low current density (about 0.05 milliampere per square foot); an uncoated structure would require high current density (about 10 milliamperes per square foot). The average current density required for cathodic protection is 2 milliamperes per square foot of *bare* area. The amount of current required for complete cathodic protection can be determined three ways:

- An actual test on existing structures using a temporary cathodic protection setup.
- A theoretical calculation based on coating efficiency.
- An estimate of current requirements using tables based on field experience.

2.1.9.1 The second and third methods above can be used on both existing and new structures.

2.1.9.2 Current requirements can be calculated based on coating efficiency and current density (current per square foot) desired. The efficiency of the coating as supplied will have a direct effect on the total current requirement, as Equation 1 shows:

$$I = (A)(I')(1.0-CE) \quad \text{(Equation 1)}$$

where I is total protective current, A is total structure surface area in square feet, I' is required current density, and CE is coating efficiency. Equation 1 may be used when a current requirement test is not possible, as on new structures, or as a check of the current requirement test on existing structures. Coating efficiency is directly affected by the type of coating used and by quality control during coating application. The importance of coating efficiency is evident in the fact that a bare structure may require 100,000 times as much current as would the same structure if it were well coated.

2.1.9.3 Current requirements also can be estimated from Table 2. The table gives an estimate of current, in milliamperes per square foot, required for complete cathodic protection. That value, multiplied by the surface area of the structure to be protected (in square feet) gives the total estimated current required. Caution should be used when estimating, however, as under- or overprotection may result.

Environment	Current density (mA/sq ft)	
	AFM 88-9 ^a	Ger ^{ard} _b
Neutral soil	0.4 to 1.5	0.4 to 1.5
Well aerated neutral soil	2 to 3	2 to 3
Wetsoil	1 to 6	2.5 to 6
Highly acidic soil	3 to 15	5 to 15
Soil supporting active sulfate-reducing bacteria	6 to 42	Up to 42
Heated soil	3 to 25	5 to 25
Stationary freshwater	1 to 6	5
Moving freshwater containing dissolved oxygen	5 to 15	5 to 15
Seawater	3 to 10	5 to 25

Table 2
Typical current density requirements for cathodic protection of uncoated steel

2.1.10 Coating resistance. A coating's resistance decreases greatly with age and directly affects structure-to-electrolyte resistance for design calculations. The coating manufacturers supply coating resistance values.

2.1.11 Protective current required. By knowing the physical dimensions of the structure to be protected, the surface area can be calculated. The product of the surface area multiplied by current density obtained previously in / above gives the total current required.

2.1.12 The need for cathodic protection. For existing structures, the current requirement survey (above) will verify the need for a cathodic protection system. For new systems, standard practice is to assume a current density of at least 2 milliamperes per square foot of bare area will be needed to protect the structure. (However, local corrosion history may demand a different current density.) In addition, cathodic protection is *mandatory* for underground gas distribution lines (Department of Transportation regulations—Title 49, *Code of Federal Regulations*, Oct 1979) and for water storage tanks with a 250,000-gallon capacity or greater. Cathodic protection also is required for underground piping systems located within 10 feet of steel reinforced concrete because galvanic corrosion will occur between the steel rebar and the pipeline.

2.2 Determining type and design of cathodic protection system. When all preliminary data have been gathered and the protective current has been estimated, the design sequence can begin. The first question to ask is: which type (galvanic or impressed current) cathodic protection system is needed? Conditions at the site sometimes dictate the choice. However, when this is not clear, the criterion used most widely is based on current density required and soil resistivity. If the soil resistivity is low (less than 5000 ohm-centimeters) and the current density requirement is low (less than 1 milliampere per square foot), a galvanic system can be used. However, if the soil resistivity and/or current density requirement exceed the above values, an impressed current system should be used. Figure 2 will be used in the design sequence. Design sequences for each type of cathodic protection system are given below.

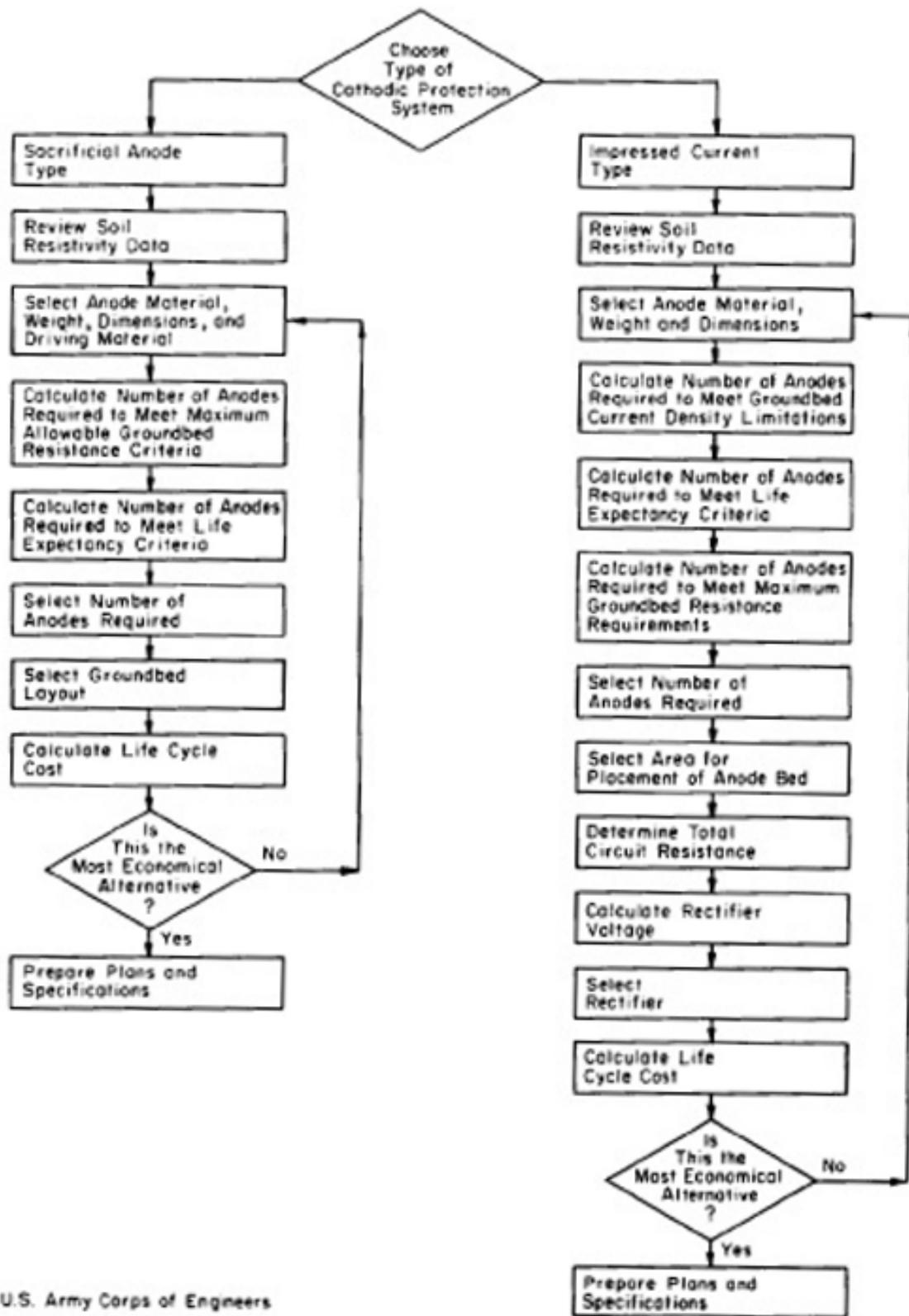


Figure 2

2.2.1 Sacrificial anode (galvanic) cathodic protection system design. The following nine steps are required when designing galvanic cathodic protection systems.

2.2.1.1 Review soil resistivity. The site of lowest resistivity will likely be used for anode location to minimize anode-to-electrolyte resistivity. In addition, if resistivity variations are not significant, the average resistivity will be used for design calculations.

2.2.1.2 Select anode. As indicated above, galvanic anodes are usually either magnesium or zinc. Zinc anodes are used in extremely corrosive soil (resistivity below 2000 ohm-centimeters). Data from commercially available anodes must be reviewed. Each anode specification will include anode weight, anode dimensions, and package dimensions (anode plus backfill), as Table 3 shows for magnesium-alloy anodes. In addition, the anode's driving potential must be considered. The choice of anode from those available is arbitrary; design calculations will be made for several available anodes, and the most economical one will be chosen.

Weight (lb)	Size (in.)	Packaged wt (lb)	Packaged size (in.)
3	3.75 x 3.75 x 5	12	6 x 10
5	3.75 x 3.75 x 7.5	17	6 x 12
9	2.75 x 2.75 x 26	35	6 x 31
9	3.75 x 3.75 x 13.25	27	6 x 17
12	3.75 x 3.75 x 18	36	6 x 23
14	2.75 x 2.75 x 41	50	6 x 46
14	3.75 x 3.75 x 21	42	6.5 x 26
17	2.75 x 2.75 x 50	60	6 x 55
17	3.75 x 3.75 x 26	45	6.5 x 29
20	2.5 x 2.5 x 59.25	70	5 x 66
24	4.5 x 4.5 x 23	60	7 x 30
32	5.5 x 5.5 x 21	74	8 x 28
40	3.75 x 3.75 x 59.25	105	6.5 x 66
48	5.5 x 5.5 x 30	100	8 x 38
48	8 x 16	100	12 x 25
60	4.5 x 4.5 x 60	—	—

Note: Core material is a galvanized 20-gage perforated steel strip. Anodes longer than 24 inches have a 9-gage core. The connecting wire is a 10-foot length of solid No. 12 AWG TW insulated copper wire, silver-soldered to the core with joints sealed against moisture. Special wires or other lengths are available.

Table 3

Weights and dimensions of selected high-potential magnesium-alloy anodes for use in soil or water

2.2.1.3 Calculate net driving potential for anodes. The open-circuit potential of standard alloy magnesium anodes is approximately -1.55 volts to a copper-copper sulfate half-cell. The open-circuit potential of high-manganese magnesium anodes is approximately -1.75 volts to a copper-copper sulfate half-cell.

2.2.1.3.1 The potential of iron in contact with soil or water usually ranges around -0.55 volt relative to copper-copper sulfate. When cathodic protection is applied using magnesium anodes, the iron potential assumes some value between -0.55 and - 1.0 volt, depending on the degree of protection provided. In highly corrosive soils or waters, the natural potential of iron may be as high as -0.82 volt relative to copper-copper sulfate. From this, it is evident that -0.55 volt should not be used to calculate the net driving potential available from magnesium anodes.

2.2.1.3.2 A more practical approach is to consider iron polarized to -0.85 volt. On this basis, standard alloy magnesium anodes have a driving potential of 0.70 volt (1.55-0.85 0.70) and high potential magnesium anodes have a driving potential of 0.90 volt (1.75-0.85 0.90). For cathodic protection design that involves magnesium anodes, these potentials, 0.70 and 0.90 volt, should be used, depending on the alloy selected.

2.2.1.4 Calculate number of anodes needed to meet groundbed resistance limitations. The total resistance (R_T) of the galvanic circuit is given by Equation 2:

$$R_T = R_A + R_W + R_C \quad \text{(Equation 2)}$$

where R_A is the anode-to-electrolyte resistance, R_W is the anode lead wire resistance, and R_C is the structure-to-electrolyte resistance. The total resistance also can be found by using Equation 3:

$$R_T = \Delta E / I \quad \text{(Equation 3)}$$

where E is the anode's driving potential discussed above and I is the current density required to achieve cathodic protection. R_C Equation 2 can be calculated by using Equation 4:

$$R_C = R \times A \quad \text{(Equation 4)}$$

where R is the average coating resistance, in ohms per square feet, at the end of the proposed lifetime for the system (R is specified by the supplier), and A is the structure's surface area in square feet. Assuming R_W in Equation 2 is negligible, that anode-to-electrolyte resistance can then be calculated from Equation 5:

$$R_a = R_T - R_C \quad \text{(Equation 5)}$$

Which gives the maximum allowable grounded resistance; this will dictate the minimum number of anodes required (as number of anodes decreases, grounded resistance increases). To calculate the number of anodes required, Equation 6 is used:

$$N = \frac{(0.0052)(\rho)}{(R_a)(L)} \left[\ln \frac{8L}{d} - 1 \right] \quad \text{(Equation 6)}$$

where N is the number of anodes, ρ is the soil resistivity in ohms, R_a is the maximum allowable grounded resistance in ohms (as computed in Equation 5), L is the length of the backfill column in feet (specified by the supplier) and d is the diameter of the backfill column in feet (specified by the supplier).

2.2.1.5 Calculate number of anodes for system's life expectancy. Each cathodic protection system will be designed to protect a structure for a given number of years. To meet this lifetime requirement, the number of anodes (N) must be calculated using Equation 7:

$$N = \frac{(L) (I)}{49.3 (W)} \quad \text{(Equation 7)}$$

Where L is the expected lifetime in years, W is weight (in pounds) of one anode, and I is the current density required to protect the structure (in milliamperes).

2.2.1.6 Select number of anodes to be used. The greater value of Equation 6 or Equation 7 will be used as the number of anodes needed for the system.

2.2.1.7 Select groundbed layout. When the required number of anodes has been calculated, the area to be protected by each anode is calculated by Equation 8:

$$A = A_T/N \quad \text{(Equation 8)}$$

Where A is the area to be protected by one anode, A_T is total surface area to be protected, and N is the total number of anodes to be used. For galvanic cathodic protection systems, the anodes should be spaced equally along the structure to be protected.

2.2.1.8 Calculate life-cycle cost for proposed design. NACE Standard RP-02 should be used to calculate the system's life-cycle cost. The design process should be done for several different anode choices to find the one with minimal life-cycle cost.

2.2.1.9 Prepare plans and specifications. When the design procedure has been done for several different anodes and the final anode has been chosen, plans and specifications can be completed.

2.2.2 Impressed current cathodic protection system design. Thirteen steps are required when designing impressed current cathodic protection systems.

2.2.2.1 Review soil resistivity. As with galvanic systems, this information will contribute to both design calculations and location of anode groundbed.

2.2.2.2 Review current requirement test. The required current will be used throughout the design calculations. The calculated current required to protect 1 square foot of bare pipe shall agree with the values in Table 2.

2.2.2.3 Select anode. As with the galvanic system, the choice of anode is arbitrary at this time; economy will determine which anode is best. Table 4 gives common anode sizes and specifications. The anodes used most often are made of high-silicon chromium-bearing cast-iron (HSCBCI). When impressed current-type cathodic protection systems are used to mitigate corrosion on an underground steel structure, the auxiliary anodes often are surrounded by a carbonaceous backfill. Backfill materials commonly used include coal coke breeze, calcined petroleum coke breeze, and natural

Anode weight (lb)	Anode dimensions (in.)	Anode surface size (in.)	Package area (sq ft)
12	1 x 60	1.4	10 x 84
44	2 x 60	2.6	10 x 84
60	2 x 60	2.8	10 x 84
110	3 x 60	4.0	10 x 84

Table 4
Weights and dimensions of selected circular high-silicon chromium-bearing cast iron anodes

graphite particles. The backfill serves three basic functions: (a) it decreases the anode-to-earth resistance by increasing the anode's effective size, (b) it extends the system's operational life by providing additional anode material, and (c) it provides a uniform environment around the anode, minimizing deleterious localized attack. The carbonaceous backfill, however, cannot be expected to increase the groundbed life expectancy unless it is well compacted around the anodes. In addition to HSCBCI anodes, the ceramic anode should be considered as a possible alternative for long-term cathodic protection of water storage tanks and underground pipes in soils with resistivities less than 5000 ohm-centimeters. The ceramic anode consumption rate is 0.0035 ounce per ampere-year compared to a 1 pound per ampere-year for HSCRCI anodes.

2.2.2.4 Calculate number of anodes needed to satisfy manufacturer's current density limitations. Impressed current anodes are supplied with a recommended maximum current density. Higher current densities will reduce anode life. To determine the number of anodes needed to meet the current density limitations, use Equation 9:

$$N = \frac{I}{(A_1 (I_1))} \quad \text{(Equation 9)}$$

where N is number of anodes required, I is total protection current in milliamperes, A1 is anode surface area in square feet per anode, and I1 is recommended maximum current density output in milliamperes.

2.2.2.5 Calculate number of anodes needed to meet design life requirement.

Equation 10 is used to find the number of anodes:

$$N = \frac{(L) (I)}{(1000) (W)} \quad \text{(Equation 10)}$$

where N is number of anodes, L is life in years, and W is weight of one anode in pounds.

2.2.2.6 Calculate number of anodes needed to meet maximum anode grounded resistance requirements.

Equation 11 is used to calculate the number of anodes required:

$$R_a = (\rho K/NL) + (\rho P/S) \quad \text{(Equation 11)}$$

where R_a is the anodes' resistance, ρ is soil resistivity in ohm-centimeters, K is the anode shape factor from Table 5, N is the number of anodes, L is length of the anode backfill column in feet, P is the paralleling factor from Table 6, and S is the center-to-center spacing between anode backfill columns in feet.

L/d	K	L/d	K
5	0.0140	20	0.0213
6	0.0150	25	0.0224
7	0.0158	30	0.0234
8	0.0165	35	0.0242
9	0.0171	40	0.0249
10	0.0177	45	0.0255
12	0.0186	.50	0.0261
14	0.0194	55	0.0266
16	0.0201	60	0.0270
28	0.0207		

Table 5

N	P	N	P
2	0.00261	14	0.00168
3	0.00289	16	0.00155
4	0.00283	18	0.00145
5	0.00268	20	0.00135
6	0.00252	22	0.00128
7	0.00237	24	0.00121
8	0.00224	26	0.00114
9	0.00212	28	0.00109
10	0.00201	30	0.00104
12	0.00182		

Table 6

2.2.2.7 Select number of anodes to be used. The highest number calculated by Equation 9, 10, or 11 will be the number of anodes used.

2.2.2.8 Select area for placement of anode bed. The area with the lowest soil resistivity will be chosen to minimize anode-to-electrolyte resistance.

2.2.2.9 Determine total circuit resistance. The total circuit resistance will be used to calculate the rectifier size needed.

2.2.2.9.1 Calculate anode groundbed resistance. Use Equation 11.

2.2.2.9.2 Calculate groundbed header cable resistance. The cable is typically supplied with a specified resistance in ohms per 100 feet. The wire resistance then is calculated from Equation 12:

$$R_w = \frac{\text{ohms (L)}}{100 \text{ ft}} \quad (\text{Equation 12})$$

where L is the structure's length in feet. Economics are important in choosing a cable, and may indeed be the controlling factor. To determine the total annual cable cost, Kelvin's Economic Law can be used as shown in Equation 13.

$$T = \frac{(0.0876)(I^2)(R)(L)(P)}{E} = (0.15)(S)(L) \quad (\text{Equation 13})$$

Where T is total annual cost in dollars per year, I is total protection current in amperes, R is cable resistance in ohms per 1000 feet, L is cable length in feet, P is cost of electrical energy in kilowatt-hour, E is the rectifier efficiency expressed as percent, and S is the cable's initial cost in dollars per foot.

2.2.2.9.3 Calculate structure-to-electrolyte resistance. Using Equation 14:

$$R_c = R \times N \quad (\text{Equation 14})$$

where R_c is the structure-to-electrolyte resistance, R is the coating resistance in ohms per square feet, and N is the coated pipe area in square feet.

2.2.2.9.4 Calculate total circuit resistance. To calculate the total resistance, R_T , equation 2-15 is used:

$$R_T = R_a + R_w + R_c \quad (\text{Equation 15})$$

2.2.2.10 Calculate rectifier voltage. Equation 16 is used to determine voltage output (V) of the rectifier:

$$V_{rec} = (I)(R_T)(150\%) \quad (\text{Equation 16})$$

where I is total protection current in amperes, R_T is total circuit resistance, and 150 percent is a factor to allow for aging of the rectifier stacks.

2.2.2.11 Select a rectifier. A rectifier must be chosen based on the results of Equation 16. Many rectifiers are available commercially; one that satisfies the minimum requirements of I and (V_{rec}) in Equation 16 should be chosen. Besides the more common rectifiers being marketed, a solar cathodic protection power supply (for d.c. power) may be considered for remote sites with no electrical power. Three factors should be considered when specifying a solar cathodic protection power supply are:

- The cost of the solar cathodic protection power supply in dollars per watt of continuous power.
- The solar cathodic protection power supply's much higher initial cost compared to selenium rectifiers operated by a.c. power.
- The additional maintenance required for a solar cathodic protection power supply, mainly to keep the solar panels free of dirt deposits.

2.2.2.12 Calculate system cost. As with the galvanic cathodic protection system, the choice of anode for design calculation is arbitrary. When several anodes have been used in the design calculations, an economic evaluation should be done as recommended in NACE Standard RP-02.

2.2.2.13 Prepare plans and specifications.

3. CURRENT REQUIREMENT TESTING

3.1 Required current. A critical element in designing galvanic and impressed current cathodic protection systems is the current required for complete cathodic protection. Complete cathodic protection is achieved when the structure potential is -0.85 volt with respect to a copper-copper sulfate reference electrode.

3.2 Sample test. Current requirement tests are done by actually applying a current using a temporary test setup, and adjusting the current from the power source until suitable protective potentials are obtained. Figure 3 shows a temporary test setup. In this setup, batteries can be used as the power supply, in series with heavy-duty adjustable resistors. The resistors can be adjusted to increase the current until the potential at the location of interest, such as point A in Figure 3, is at -0.85 volt with respect to a copper-copper sulfate reference cell. The current supplied is the current required for cathodic protection. The effectiveness of the insulating joints shown in Figure 3 can also be tested. The potentials at points B and C are measured, first with the current interruptor switch closed, then with it open. If there is any difference between the two readings at either point, the joint is not insulating completely.

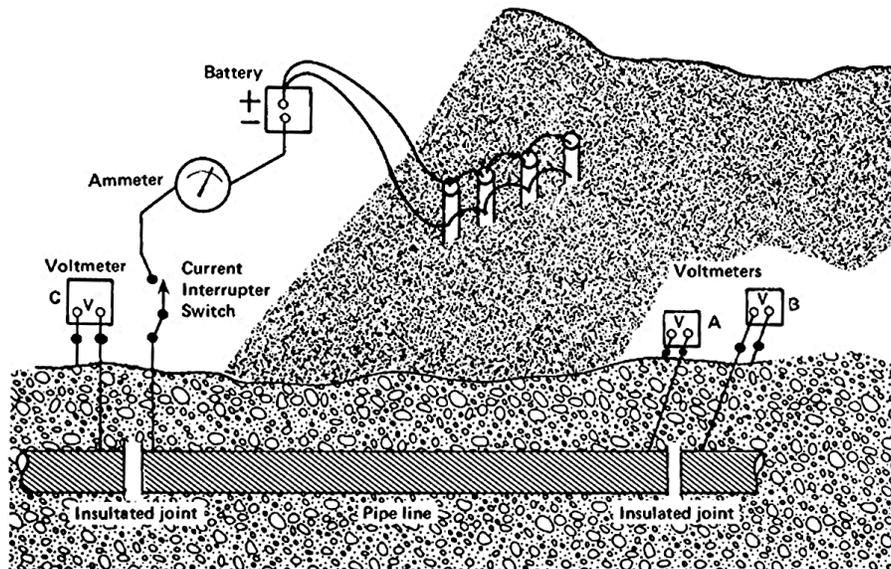


Figure 3

4. EXAMPLES OF GALVANIC CATHODIC PROTECTION DESIGN

4.1 Underground steel storage tank.

Galvanic cathodic protection is designed for an underground steel storage tank shown in figure 4-1. The tank is already installed and current requirement tests have been made.

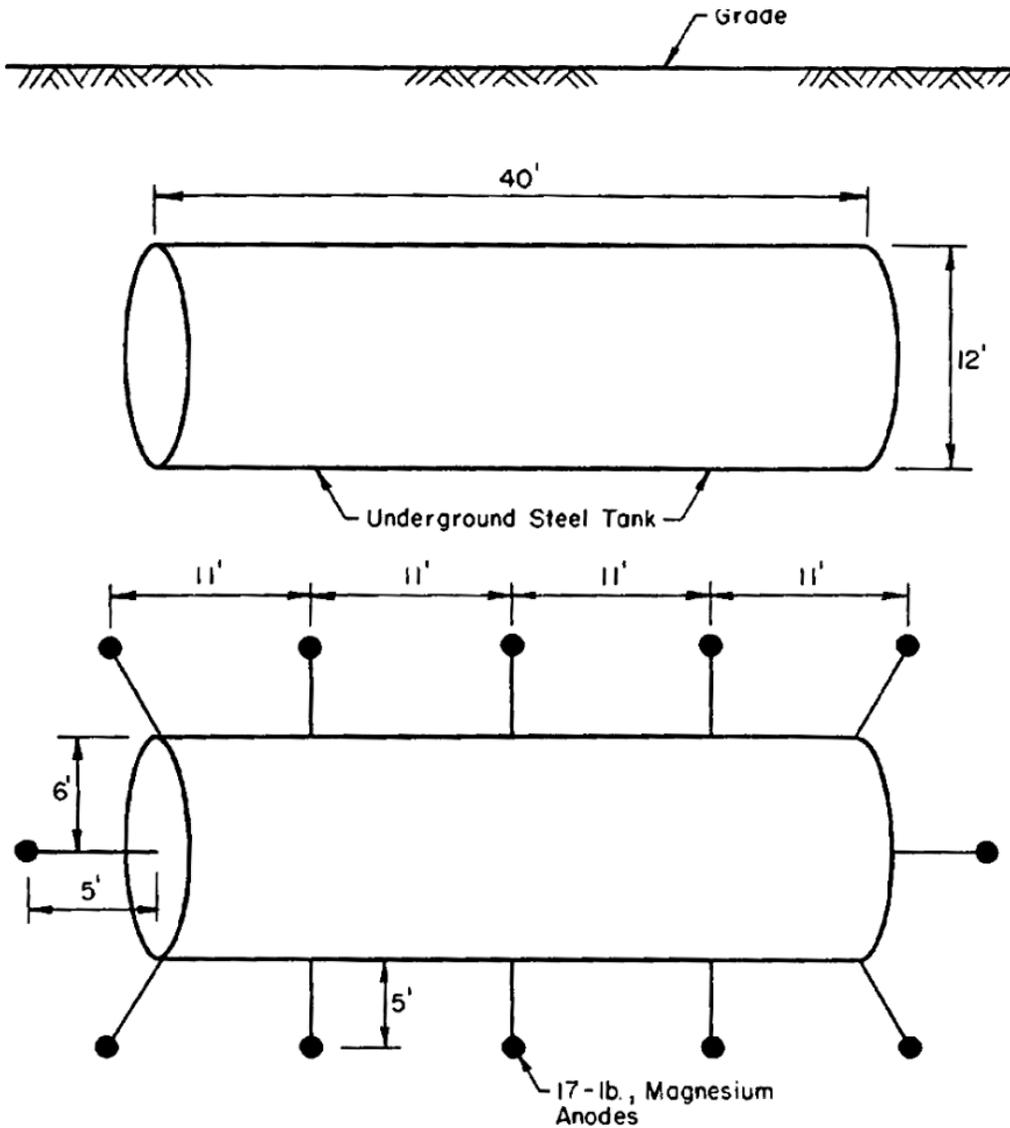


Figure 4-1

4.1.1 Design data.

- (1) Tank diameter is 12 feet.
- (2) Tank length is 40 feet.
- (3) Design for 80 percent coating efficiency, based on experience.
- (4) Design for 15-year life.
- (5) Current requirement is 0.7 ampere.
- (6) Packaged 17-pound standard magnesium anodes must be used.
- (7) The tank is insulated well enough from foreign structures.

4.1.2 Computations.

- (1) Find the minimum weight of anodes required for the tank using equation 4-1:

$$W = YSI/E \quad \text{(Equation 4-1)}$$

where $Y = 15$ years, $S = 8.8$ pounds per ampere-year, $I = 0.7$ ampere, and $E = 0.50$ efficiency. Thus,

$$W = (15)(8.8)(0.7)/0.50 = 184.5 \text{ lb}$$

- (2) Find the number of magnesium anodes (17 pounds each) required:

$$N = 184.8/17 = 10.9 \text{ (use 12 anodes for symmetry).}$$

4.1.3 Placement. Locate anodes as shown in figure 4-1.

4.2 Gas distribution system.

Galvanic cathodic protection is designed for a gas distribution system in a residential area as shown in figure 4-2.

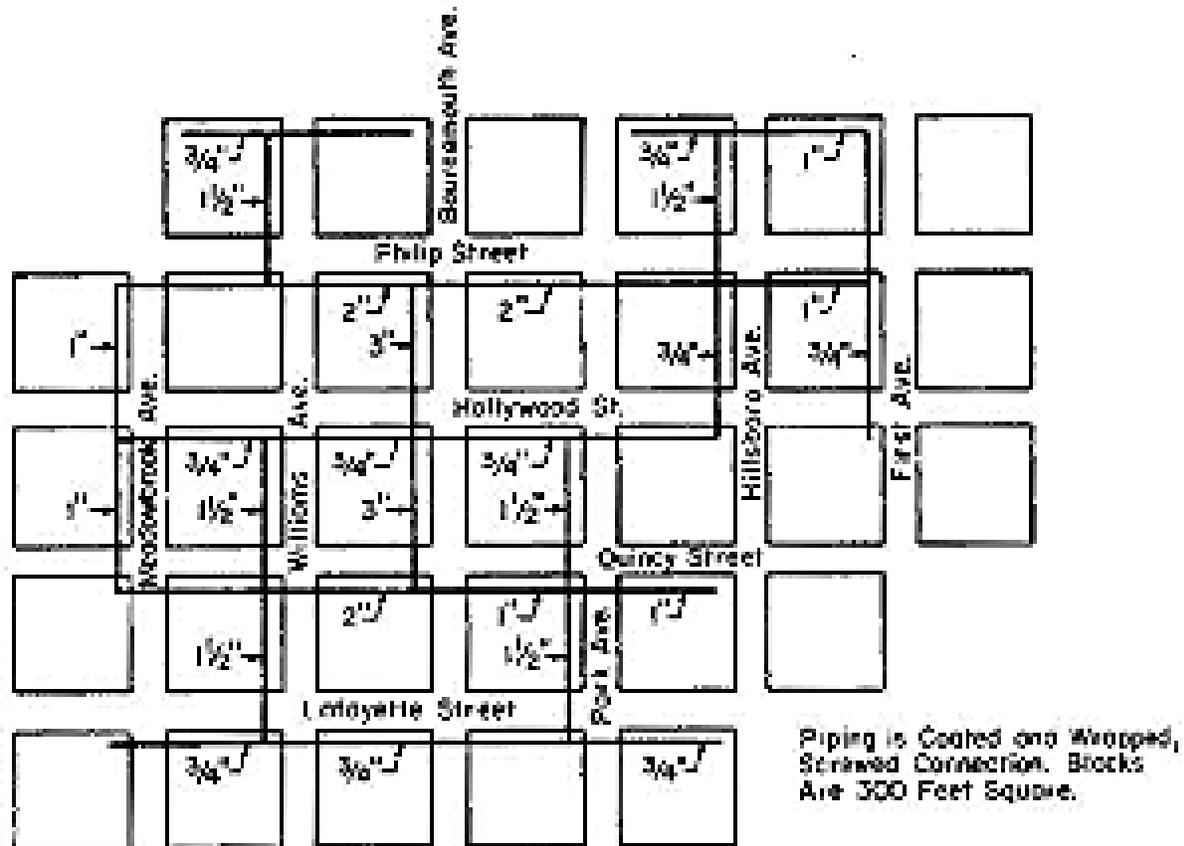


Figure 4-2

4.2.1 Design data.

- (1) Average soil resistivity is 4500 ohm-centimeters.
- (2) Design for 90 percent coating efficiency, based on experience.
- (3) Design for 15-year life.
- (4) Design for 2 milliamperes per square foot of bare pipe.
- (5) Packaged-type magnesium anodes must be used.
- (6) Insulating couplings are used on all service taps. Mains are electrically isolated from all other metal structures in the area.
- (7) All pipe has been pre-coated at the factory and wrapped with asbestos felt.
The coating has been tested over the trench for holidays and defects have been corrected. The coating is considered to be better than 99.5 percent perfect when installed.

4.2.2. Computations.

(1) Find the total outside area of piping (table 4-1).

(2) Find the area of bare pipe to be protected cathodically based on 90 percent coating efficiency:

$$A = 4288 \times 0.1$$

$$A = 429 \text{ sq. ft.}$$

Pipe size (in.)	Pipe length (ft)	Pipe area (sq ft/in ft)	Pipe area (sq ft)
3	600	0.916	550
2	1500	0.622	933
1½	1800	0.497	594
1	2400	0.344	506
	3900	0.278	1084
Total area of pipe in square feet			4288

Table 4-1

(3) Find the maximum protective current required based on 2 milliamperes per square foot of bare metal:

$$I = 2 \times 429$$

$$I = 858 \text{ mA or } 0.858 \text{ A.}$$

(4) Find the weight of anode material required based on maximum current requirement and 15-year life. Use equation 4-1:

$$W = (YSI/E)$$

Equation 4-1

where Y = 15 years, S = 8.8 pounds per ampere-year, I = 0.858 ampere, and E = 0.50 efficiency. Thus,

$$W = [(15 \text{ yr})(8.8 \text{ lb/A}\cdot\text{yr})(0.858 \text{ A})]/0.50$$

$$W = 227 \text{ lb}$$

Note that the 227-pound value is based on an output current of 0.86 ampere for the cathodic protection system's full design life, 15 years. Strictly speaking, this is not the true condition, because current output after new installation is much less due to the high coating efficiency. The average current requirement at first may be as low as 0.03 milliamperes per square foot of pipe.

(5) Find the current output to ground for a single 17-pound standard packaged magnesium anode using equation 4-2:

$$i = (Cfy)/P \quad \text{(Equation 4-2)}$$

where $C = 120,000$, a constant for well coated structures using magnesium, $f = 1.00$, $y = 1.00$, $P = 4500$ ohm-centimeters. Values for f (galvanic anode size factor) and y (structure potential factor) have been obtained from technical literature. Thus,

$$i = (120000 \times 1.00 \times 1.00)/4500 \text{ ohm-cm} = 26.7 \text{ mA.}$$

Because the structure is well coated, anode spacing will be relatively large.

(6) Find the number of anodes (n) required from equation 4-3:

$$n = I/i \quad \text{(Equation 4-3)}$$

where $I = 858$ milliamperes and $i = 26.7$ milliamperes.

$$n = 858/26.7 = 32.1$$

Thus, $n = 32.1$ (use 32 anodes).

(7) Find the anode distribution.

(a) Pipe area protected by one anode:

$$A = 4288/32 = 134 \text{ sq ft/anode}$$

(b) Find the anode division (table 4-2).

Pipe size (in.)	Pipe area (sq ft)	Pipe length (ft)	Number of anodes	Anode spacing (ft)
3	550	600	4	150
2	933	1500	7	214
1½	895	1800	7	257
1	826	2400	6	400
¾	1084	2900	<u>8</u>	488
Total number of anodes			32	

Table 4-2

Dimensions for finding anode division

5. EXAMPLES OF IMPRESSED CURRENT CATHODIC PROTECTION DESIGN

5.1 Steel gas main.

Impressed current cathodic protection is designed for the 6-inch welded gas main shown in figure 5-1. This pipeline is not yet constructed, so measurements cannot be taken.

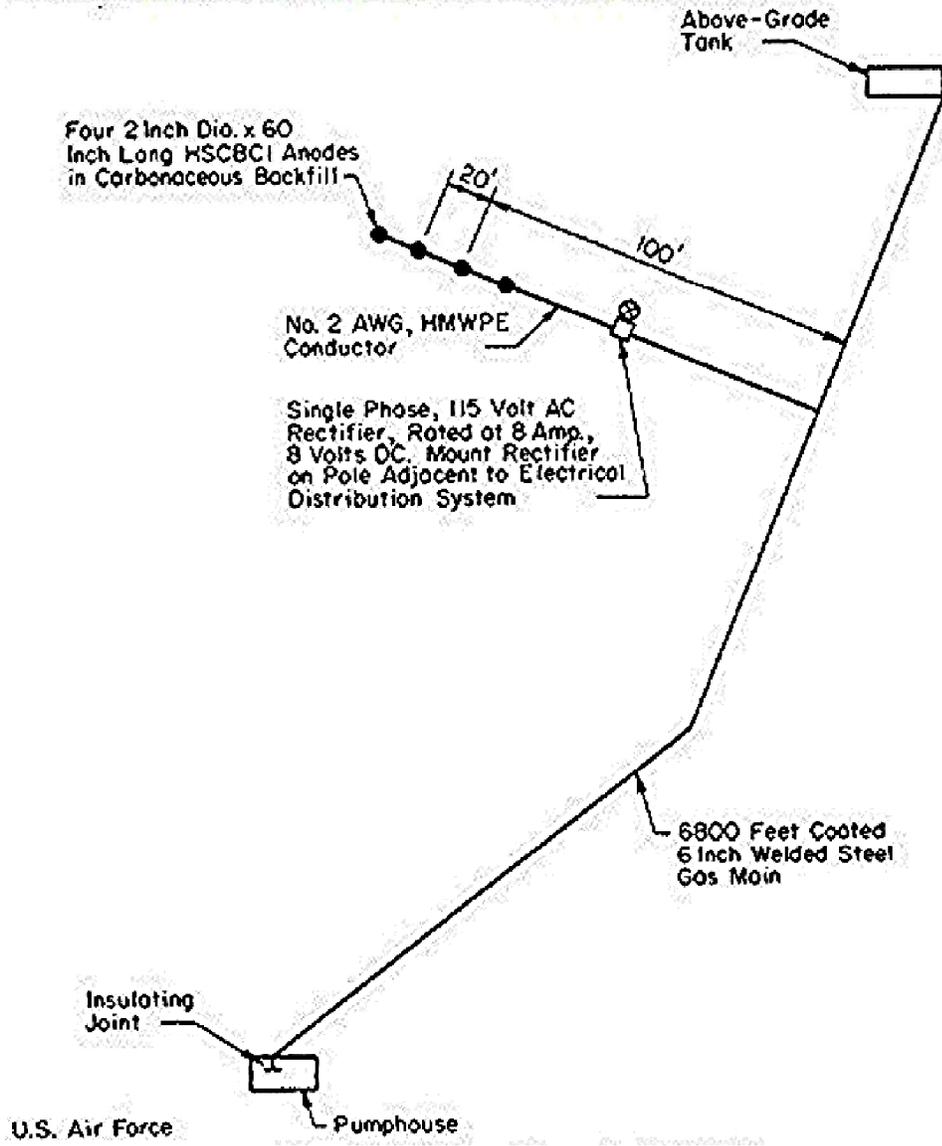


Figure 5-1

5.1.1 Design data.

- (1) Average soil resistivity is 2000 ohm-centimeters.
- (2) Effective coating resistance at 15 years is estimated at 2500 ohms per square foot.
- (3) Pipe has a 6-inch outside diameter.
- (4) Pipe length is 6800 feet.
- (5) Design for 15-year life.
- (6) Design for 2 milliamperes per square foot of bare pipe.
- (7) Design for 90 percent coating efficiency based on experience.
- (8) The pipeline must be isolated from the pumphouse with an insulating joint on the main line inside the pumphouse.
- (9) HSCBCI anodes must be used with carbonaceous backfill.
- (10) The pipe will be coated with hot-applied coal-tar enamel and will be holiday-checked before installation.
- (11) Anode bed must not exceed 2 ohms.
- (12) Electric power is available at 120/240 volts a.c. single phase from a nearby overhead distribution system.
- (13) Current requirement test indicates that 2.36 amperes are needed for adequate cathodic protection.

5.1.2 Computations.

- (1) Find the gas main's outside area:

Pipe size - 6 in.

Pipe length - 6800 ft

Pipe area – $(6800)([\pi/2](A) = (L)(\pi)(d) = (6800)(\pi)(6/12) = 10,681$ sq ft.

- (2) Check the current requirement:

$$I = (A)(I')(1.0 - CE)$$

$$I = 10681 \text{ sq ft} (2 \text{ mA/sq ft})(1.0 - 0.9)$$

$$I = 2136 \text{ mA},$$

which agrees with the current requirement test indicated in (13) above.

(3) Select an anode from manufacturer's literature (high-silicon chromium-bearing cast-iron anodes). Choose a 60-pound anode with a 2.8-square-foot surface area (arbitrary selection).

(4) Calculate the number of anodes needed to meet the anode supplier's current density limitations:

$$N = I/(A_1)(I_1) = 2360 \text{ mA}/(28 \text{ sf/anode})(1000 \text{ mA/sf}) = 0.84 \text{ anode}$$

(Recommended maximum current density output for high-silicon chromium-bearing cast-iron anodes is 1000 mA/sf.)

(5) Calculate the number of anodes required to meet the design life requirements:

$$N = (L)(I)/(1000)(W) = (15 \text{ yrs})(2360 \text{ mA})/(1000)(60 \text{ lb/anode}) = 0.59 \text{ anode}$$

(6) Calculate the number of anodes required to meet maximum anode groundbed resistance requirements:

$$R_a = (\rho K/LN) + (\rho P/S)$$

$$N = \rho K/(L(R_a - (\rho P/S))) = 2.75 = 3 \text{ anodes}$$

(7) Select the number of anodes to be used. Since the last calculation resulted in the largest number of anodes, it will be used. The groundbed resistance, R_a , using three anodes, would equal 1.86 ohms; to insure compliance with the manufacturer's limitations, four anodes will be used.

(8) Select an area for anode bed placement. The area of lowest resistivity will be used, which is 100 feet from the pipeline.

(9) Determine the total circuit resistance.

(a) Calculate the anode groundbed resistance:

$$R_a = (\rho K/LN) + (\rho P/S) = 1.46 \text{ ohms}$$

(b) Calculate the groundbed resistance for a 50-foot header cable using equation 2-12. The resistance specified by the manufacturer is 0.0159 ohm per 100 ft of No.2 AWG cable:

$$R_w = (\text{ohms/ft})(L)$$

$$R_w = (0.0159 \text{ ohm/100 ft})(500 \text{ ft}) = 0.0795 \text{ ohm}$$

(c) Calculate the structure-to-electrolyte resistance:

$$R_c = R \times N = (2500 \text{ ohms/sf}) \times (11,800 \text{ sf}) = 0.212 \text{ ohm}$$

(d) Calculate the total resistance (eq 2-15):

$$R_T = R_a + R_w + R_c = 1.46 \text{ ohm} + 0.0795 \text{ ohm} + 0.212 \text{ ohm} = 1.75 \text{ ohms}$$

(10) Calculate the rectifier voltage from equation 2-16:

$$V_{\text{rec}} = (I)(R_T)(150\%) = (2.36 \text{ A})(1.75 \text{ ohms})(150\%) = 6.2 \text{ V}_{\text{rec}}$$

(11) Select rectifier. Based on the design requirement of 6.2 volts and 2.36 amperes, a rectifier can be chosen from those marketed. After a rectifier has been chosen, the

system's cost can be calculated. A comparison with other anode sizes and types will yield the most economical design.

5.2 Heating distribution system.

Impressed current cathodic protection is designed for a well coated, buried heating distribution system as shown in figure 5-2. The distribution system has not yet been installed, so measurements cannot be made. Rectifier size need not be calculated, because it is sized in the field after anode installation.

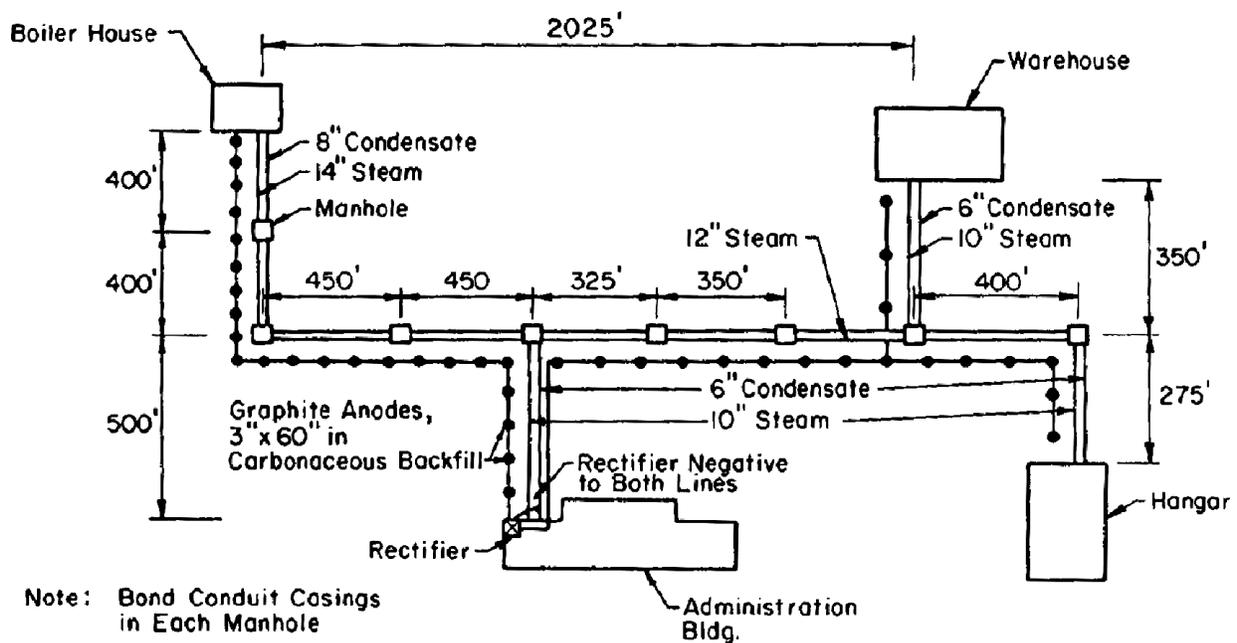


Figure 5-2
Heating Distribution System

5.2.1 Design data.

- (1) Average soil resistivity is 1000 ohm-centimeters.
- (2) Design for 80 percent coating efficiency based on experience.
- (3) Design for 4 milliamperes per square foot of bare metal heating conduits.
- (4) Groundbed resistance must not exceed 1.5 ohms.
- (5) Graphite anodes must be installed with carbonaceous backfill.

- (6) Design for a 15-year life.
- (7) Insulating joints must be provided on both steam and condensate lines at the first flange connection inside all buildings.
- (8) All conduit must be metal-bonded together in each manhole.
- (9) All conduit will be precoated at the factory and will not have been holiday-checked.
- (10) Single-phase electrical power is available at 120/240 volts a.c. from the administration building.

5.2.2 Computations.

- (1) Find the conduit's total outside area. Because the gage of the metal from which the conduit is made ranges between 14 and 16, the pipe's outside diameter is considered the same as the inside diameter.

- (a) Steam conduit area must be calculated (table 5-1).

Conduit size (in.)	Conduit length (ft)	Conduit area (sq ft/ (lin ft)	Conduit area (sq ft)
14	1700	3.67	6239
12	1125	3.14	3533
10	1525	2.62	3996
Total area of steam conduit		13.768	

Table 5-1

Dimensions for finding steam conduit area: heat distribution system

- (b) Condensate return conduit area must be calculated (table 5-2).

Conduit size (in.)	Conduit length (ft)	Conduit area (sq ft/ lin ft)	Conduit area (sq ft)
8	1700	2.09	3553
6	2650	1.57	<u>4161</u>
Total area of condensate return conduit			7713
Total outside area of all conduit			21481

Table 5-2

Dimensions for finding condensate return conduit area: heat distribution system

(2) Find the area of bare pipe to be cathodically protected based on 80 percent coating efficiency:

$$A = 21,481 \times 0.2$$

$$A = 4296 \text{ sq ft.}$$

(3) Find the maximum protective current required based on 4 milliamperes per square foot of bare metal:

$$I = 4296 \times 4$$

$$I = 17,184 \text{ mA or } 17.2 \text{ A.}$$

(4) Compute the maximum weight of anode material needed for 15 years' life.

(a) Graphite anodes are used.

(b) Average deterioration rate for graphite is 2.0 pounds per ampere-year.

(c) Find the maximum weight of anode material required:

$$W = YSI/E$$

Where Y = 15 years, S = 2.0 lb/ampere-year, I = 17.2 amperes, and E = 0.50 efficiency. Thus:

$$W = (15)(2.0)(17.2)/(0.50) = 1032 \text{ lb}$$

5.2.3 Groundbed Design

- (1) Anode size is 3" x 60" (backfilled (10" x 84")) and weight is 25 lb per node unit.
- (2) Find the resistance to earth of a single anode:

$$R_v = PK/L$$

Where P = 1000 ohm-cm, L = 7.0 ft (backfilled size) and K = 0.0167, L/d = 8.4 (manufacturer's data). Thus:

$$R_v = (1000)(0.0167)/(7.0) = 2.39 \text{ ohms}$$

- (3) Compute the number of anodes required. The low resistance (2.39 ohms) of a single anode and the heavy weight of anode material required (1032 pounds) for a 15-year life indicate that the controlling factor is the amount of anode material, not groundbed resistance. The minimum number of anodes (N) required is:

$$N = 1032/25 = 41.3 \text{ or } 41 \text{ anodes.}$$

These are arranged in a distributed groundbed as shown in figure D-2 based on the following estimates.

- (4) Anode distribution:

- (a) Conduit area in sections 1 through 6 of figure 5-2 are given in table 5-3.

Section	Length (ft)	Surface area (sq ft)
1	1700	$3553 + 6239 = 9792$
2	500	$785 \times 1310 = 2095$
3	1125	$1766 \times 3533 = 5299$
4	350	$550 + 917 = 1467$
5	400	$628 + 1048 = 1676$
6	275	$432 + 721 = 1153$

Table 5-3
Conduit area: heat distribution system

(b) The area of conduit protected by one anode is —

$$A = 21,481/41$$

$$A = 524 \text{ sq ft/anode.}$$

(c) Anodes will be divided as shown in table 5-4.

Section	Surface area/ anode protective area	Number of anodes
1	$9792/524 =$	19
2	$2095/524 =$	4
3	$5299/524 =$	10
4	$1467/524 =$	3
5	$1676/524 =$	3
6	$1153/524 =$	2

Table 5-4
Anode division: heat distribution system

(d) Rectifier location. Locate the rectifier in front of the administration building as figure 5-2 shows. The rectifier will be sized after anodes are installed.

QUIZ

AN INTRODUCTION TO CATHODIC PROTECTION

1. The combination of an anode, cathode, electrolyte and metallic connection is called a:
 - a. Potentiometer
 - b. Catalyst
 - c. Corrosimeter
 - d. Corrosion cell
2. A galvanic cathode protection system is called a:
 - a. Sacrificial anode system
 - b. Impressed current system
 - c. Express potential system
 - d. Two-cell system
3. The main difference between galvanic and impressed current systems is the:
 - a. galvanic system relies on alternating current (a.c.) power
 - b. impressed current system uses an external power source
 - c. galvanic system requires a protective coating system for the protected structure
 - d. impressed current systems uses a potentiometer
4. The soil resistivity range for "mild" corrosivity is:
 - a. 0 – 2000 ohm-cm
 - b. 2000 – 10000 ohm-cm
 - c. 10000 – 30000 ohm-cm
 - d. above 30000 ohm-cm
5. The typical range of current density (in mA/sq ft) required for cathodic protection of an uncoated steel tank in well aerated neutral soil is:
 - a. 0.4 to 1.5
 - b. 2 to 3
 - c. 1 to 6
 - d. 3 to 15

6. In the design of a cathodic protection system for a new underground steel pipeline, standard practice is to assume a current density of at least _____ milliamperes per square foot of bare pipe area will be needed for its protection.

- a. 0.25
- b. 0.5
- c. 1
- d. 2

7. If soil resistivity is less than 5000 ohm-centimeters and current density is less than 1 milliamperes per square foot, which of the following the most likely cathodic protection system to be warranted?

- a. sacrificial anode
- b. impressed current
- c. cathode-anode couple
- d. any of the above

8. There are _____ steps in designing a sacrificial anode cathodic protection system.

- a. 6
- b. 7
- c. 8
- d. 9

9. There are _____ steps in designing an impressed current cathodic protection system.

- a. 10
- b. 11
- c. 12
- d. 13

10. The open-circuit potential of standard alloy magnesium anodes to a copper-copper sulfate half-cell is approximately:

- a. - 1.45 volts
- b. - 1.55 volts
- c. - 1.65 volts
- d. - 1.75 volts

11. The open-circuit potential of high-manganese magnesium anodes to a copper-copper sulfate half-cell is approximately:
- 1.45 volts
 - 1.55 volts
 - 1.65 volts
 - 1.75 volts
12. Which of the following is not a step in the design of a sacrificial anode cathodic protection system?
- Select number of anodes
 - Select rectifier
 - Select number of anodes
 - Select grounded layout
13. In a sacrificial anode cathodic protection system, galvanic anodes usually either:
- copper or zinc
 - zinc or magnesium
 - magnesium or copper
 - any of the above
14. In a sacrificial anode cathodic protection system, zinc anodes are usually used in soil with soil resistivity (ohm-cm) in the range of:
- 0 to 2000
 - 2000 to 10,000
 - 10,000 to 30,000
 - 30,000 to 100,000
15. A protective coating's resistance to corrosion of a buried steel tank can be expected to:
- remain constant long-term (> 30 years)
 - decrease greatly with age
 - decrease slightly with age
 - impossible to project
16. Corrosion is an electrochemical process in which a current leaves a structure at the anode site, passes through an electrolyte, and re-enters the structure at the _____ .
- cathode
 - ground rod
 - button anode
 - rectifier

17. Impressed current cathodic protection systems use the same elements as the galvanic protection system, only the structure is protected by applying a current to it from a/an _____.

- a. cathode
- b. anode
- c. electrolyte
- d. ground wire

18. All _____ must be eliminated from existing and new cathodic protection systems.

- a. rectifiers
- b. electrolytes
- c. short circuits
- d. galvanic cells

19. A structure's corrosion rate is proportional to the _____ resistivity.

- a. coating
- b. ground
- c. anode
- d. electrolyte

20. Studying the corrosion history in the area can prove very _____ when designing a cathodic protection system.

- a. helpful
- b. misleading
- c. difficult
- d. easy

21. In general, steel's corrosion rate increases as _____ decreases when soil resistivity remains constant.

- a. current
- b. voltage
- c. pH
- d. AIC

22. galvanic anodes are usually either _____ or zinc.

- a. iron
- b. carbon
- c. magnesium
- d. manganese

23. A cathodic protection system will be designed to protect a structure _____.

- a. permanently
- b. for 5 years
- c. for 25 years
- d. for a given number of years

24. A critical part of design calculations for cathodic protection systems on existing structures is the amount of current required per square foot (called *current density*) to change the structure's potential to _____ volt.

- a. -0.85
- b. +0.85
- c. -1.85
- d. +1.85

25. The anodes used most often in impressed current cathodic protection systems are made of _____.

- a. high-sodium carbon-bearing cast-iron (HSCBCI)
- b. high-silicon chromium-bearing cast-iron (HSCBCI)
- c. high-silicon carbon-bearing cast-iron (HSCBCI)
- d. high-sodium chromium-bearing cast-iron (HSCBCI)